

DoE-Assisted Development of a 2H-MoS₂-Catalyzed Approach for the Production of Indole Derivatives

Marc Morant-Giner¹, Giuseppe Gentile², Lucia Cardo³, Michele Melchionna², Paolo Fornasiero^{2,4}, Maurizio Prato^{2,3,5}, Giacomo Filippini²

¹ Instituto de Ciencia Molecular (ICMol), Universitat de València, c/Catedrático José Beltrán 2, 46980, Paterna, Spain

² Department of Chemical and Pharmaceutical Sciences, INSTM UdR Trieste, University of Trieste, via Licio Giorgieri 1, 34127, Trieste, Italy

³ Centre for Cooperative Research in Biomaterials (CIC BiomaGUNE), Basque Research and Technology Alliance (BRTA), Paseo de Miramón 194, 20014, Donostia-San Sebastián, Spain

⁴ Istituto di Chimica dei Composti Organometallici – Consiglio Nazionale delle Ricerche (ICCOM-CNR), via Licio Giorgieri 1, 34127, Trieste, Italy

⁵ Institution Basque Foundation for Science Ikerbasque, Plaza Euskadi 5, 48013, Bilbao, Spain

marc.morant@uv.es

2H-MoS₂ is an appealing semiconductor material due to its earth-abundant nature, cheapness, and low toxicity.[1] Although 2H-MoS₂ has shown promising catalytic activity for various energy-related processes, its use in the catalysis of C-C bond forming reactions is still largely unexplored. The lack of examples in organic synthesis is mostly a consequence of the intrinsic difficulties of using bulk 2H-MoS₂ (e.g., low surface area), which implies the use of high catalytic loadings.[2] In our research, we have focused on the development of a 2H-MoS₂-mediated synthesis of valuable bisindolylmethane derivatives (**3**), using indoles (**1**) and benzaldehydes (**2**) as starting materials. Notably, with the aid of Design of Experiments (DoE) method, we have effectively established the optimal reaction conditions while also identifying the critical parameters affecting the catalytic performance of commercial 2H-MoS₂ powder (**Fig. 1**). Lastly, we have demonstrated that the catalytic system has large versatility and good tolerance towards functional group variations of the reagents.

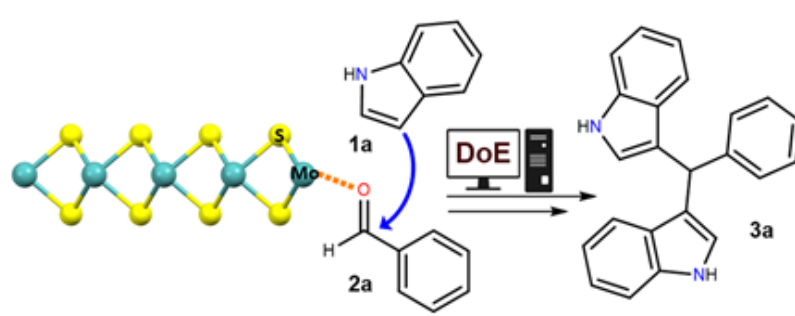


Figure 1. Schematic representation of the reaction between **1a** and **2a** to give **3a** catalysed by bulk 2H-MoS₂. For simplicity, bulk MoS₂ has been represented as a single layer.

References

[1] Song, I.; Park, C.; Choi, H. C. Synthesis and properties of molybdenum disulphide: from bulk to atomic layers. *RSC Adv.* **2014**, *5* (10), 7495-7514.

[2] Bahuguna, A.; Kumar, S.; Sharma, V.; Reddy K. L.; Bhattacharyya, K.; Ravikumar, P. C.; Krishnan, V. Nanocomposite of MoS₂-RGO as Facile, Heterogeneous, Recyclable, and Highly Efficient Green Catalyst for One-Pot Synthesis of Indole Alkaloids. *ACS Sustainable Chem. Eng.* **2017**, *5* (10), 8551-8567.